Total Synthesis of U-71, 184, A Potent New Antitumor Agent Modeled on CC-1065

by: M.A. Warpehoski*, Cancer and Viral Diseases Research, The Upjohn Company, Kalamazoo, MI 49001

<u>Abstract</u>: The synthesis of U-71,184 (2), a highly potent analog of the novel antitumor antibiotic CC-1065, is described, the penultimate step of which involves the unmasking of a <u>p</u>-hydroxy phenethyl mesylate, which undergoes facile intramolecular elimination to afford the reactive cyclopropylspirocyclohexadienone. Its enantiomer, U-71,185, was also prepared and shown to be biologically inactive.

CC-1065 (1) is one of the most potent antitumor antibiotics known.¹ It possess a unique DNA alkylating capability in its cyclopropylpyrroloindole "left-hand segment,"² and an extraordinary affinity and specificity for binding to the minor groove of B-DNA, apparently attributable in large measure to its DNA groove-complementary shape.^{1,2} The novel structural features of its component pyrroloindole segments has stimulated a great deal of recent synthetic interest in CC-1065.³ Since the publication of the first synthesis of the biologically active but much less potent left-hand segment,^{3a} **3**, our own efforts have been motivated by the search for potent and active analogs of CC-1065, which might be devoid of the delayed toxicity that precludes the clinical development of the natural product.⁴

We wish to report here the synthesis of U-71,184 (2) a new antitumor agent modeled on CC-1065 with greatly superior activity against a variety of murine tumors and human tumor xenografts.⁵ Like CC-1065, U-71,184 binds to double stranded B-DNA, and is equipotent with the natural product both <u>in vitro</u> and <u>in vivo</u>, but unlike CC-1065, it does not cause delayed death in mice.⁵

U-71,184 may be viewed as a simplified model of CC-1065, possessing its essential electronic features and general shape and dimensions, but synthetically much more accessible, given an adaptable left-hand segment synthesis and a viable strategy for convergence with a right-hand appendage.



We employed the Wierenga strategy,^{3a} with the modifications discussed below, for the construction of the key dihydropyrrolo[3,2-e]indole intermediate 4, which was then resolved through its diasteriomeric esters with N-BOC-L-tryptophan, to afford 4R and 4S (Scheme 1). *Scheme 1*



i. H₂, PtO₂; *ii*. MsCl, pyr., 0°; *iii*. NaOAc, DMF, EtOH, Δ; *iv*. HNO₃, CH₂NO₂, 0°; *v*. 1) SO₂Cl₂, **9**, Proton Sponge, CH₂Cl₂, -78°; *2*) Et₃N; *3*) HOAc (for **9**c); *vi*. BH₃ SMe₂, THF, mol-sieves; *vii*. N-BOC-L-tryptophan, EDC, CH₂Cl₂; separation *viii*. NaOH

In the original synthesis^{3a} **5a** was converted to **5b**, before reduction to, presumably, **6b**, and spontaneous cyclization to the indoline **7b**, which was quickly mesylated. This route often gave poor yields on scale-up, presumably because of intermolecular reactions of **6b** and **7b** in concentrated solutions. To circumvent this problem diol **5a** was directly hydrogenated to the anilino diol **6a** (75%), which was mesylated and cyclized in situ to **7a** (90%). Acetyl exchange, nitration, and reduction^{3a} afforded **8** (40% from **7a**).

The most problematic step in the Wierenga synthesis was the elaboration of the 8-methylindole ring of 4, which was carried out elegantly, but in a disappointingly low 20% yield, <u>via</u> Gassman's oxindole synthesis⁶ from aniline 8 and sulfide 9a, followed by efficient reduction of the resulting 10a to 4 with borane-dimethylsulfide.⁷ Model studies (Scheme 2) on *o*-anisidine (11a) and 5-methyl-*o*-anisidine (11b) indicated that a substituent adjacent to the ring position attacked by the Scheme 2



intermediate azasulfonium ylide (12) presented a steric impediment that significantly reduced the yields in this reaction.⁸ Consistent with this argument was our observation that the less sterically demanding sulfide 9c (Scheme 1) afforded the corresponding oxindole upon reaction with 8 in 85-90% yield. While this oxindole could be α -methylated with Na₂CO₃/CH₃I to afford, after careful purification from N-methylated side products, 10a (60%), a shorter and cleaner alternative was the use of the cyclic sulfide 9b⁹ in the Gassman reaction with 8. The relief of steric crowding occasioned by tying back the substituents on the sulfide into a ring system was reflected in the 58% yield of 10b, directly from 8. Borane-dimethyl-sulfide smoothly reduced 10b to racemic 4 (81%).⁷

Numerous attempts to resolve 4 by its esterification with chiral acids, including (+) MTPA, were unsuccessful. Finally (Scheme 1) derivatization of 4 with N-BOC-L-tryptophan resulted in an extraordinarily selective crystallization of the **S**,**S** diasteriomer. A single additional recrystallization (THF, hexane) afforded material of >99% diasteriomeric purity, judged by analytical HPLC, in 58% of the theoretical yield. Saponification gave 4S (m.p. 165-166°C [α]_D²⁵ -30, c = 0.58, EtOH). The mother liquors were carefully chromatographed (silica gel, THF/hexane) to provide a sample, after saponification, of 4R in comparable optical purity (m.p. 163-165°C [α]_D²⁵ + 29, c = .82, EtOH).¹⁰



i. Red-Al (6 eq.), THF, toluene, 85°, 0.5 hr; *ii.* EDC (1 eq.), DMF, 25°, 48 hr; *iii.* MsCl (4 eq.), pyridine, DMAP, 25°, 0.5 hr; *iv.* Nal (7 eq.), TMS-Cl (7 eq.), CH₃CN, PhCN, 80°, 0.5 hr; *v.* Et₃N (excess), THF, EtOAc, 25°, 0.5 hr

The convergence of the left and right-hand segment precursors and their elaboration into U-71,184 is shown in Scheme 3. As anticipated by Wierenga,^{3a} Red-Al removal of the sulfonamide protecting group readily afforded the air sensitive indoline 14, which, was condensed with 15^{11} to give the poorly soluble adduct 16 as a crude solid (70% from 4). Activation of the homobenzylic alcohol was effected with MsCl to give 17 nearly quantitatively. Initial efforts to remove the benzyl protecting group from 17 by catalytic hydrogenation yielded only 5% (from 16) of 2. In situ generated TMS-I (6-8 eq) in refluxing CH₃CN was more successful. The crude product mixture, containing both 18 and 2, was reacted with triethylamine to effect complete ring closure. The

chromatographed product (2, U-71, 184) precipitated from a small volume of acetone containing 1% triethylamine as a light yellow powdery solid (20-30% from 16).¹²

In like manner **4R** was converted to U-71,185, the enantiomer of U-71,184. U-71,185 was two orders of magnitude less cytotoxic to L1210 cells <u>in vitro</u> than was U-71,184, and was inactive <u>in vivo</u> in systems and at doses for which U-71,184 was highly curative.¹³

References

- a) C.G. Chidester, W.C. Krueger, S.A. Mizsak, D.J. Duchamp, and D.G. Martin, J. Amer. Chem. Soc. <u>103</u>, 7629 (1981) and references therein. b) L.H. Li, D.H. Swenson, S. Schpok, S.L. Kuentzel, B.D. Dayton, and W.C. Krueger, Cancer Res. <u>42</u>, 999 (1982). c) D.H. Swenson, L.H. Li, L.H. Hurley, J.S. Rokem, G.L. Petzold, B.D. Dayton, T.L. Wallace, A.H. Lin, and W.C. Krueger, Cancer Res. <u>42</u>, 2821 (1982). d) B.K. Bhuyan, K.A. Newell, S.L. Crampton, and D.D. VonHoff, Cancer Res. <u>42</u>, 3532 (1982). e) Review: V.L. Reynolds, J.P. McGovren, and L.H. Hurley, J. Antibiot. <u>39</u>, 319 (1986).
- a) D.R. Needham-VanDevanter, L.H. Hurley, V.L. Reynolds, N.Y. Theriault, W.C. Krueger and W. Wierenga, Nucleic Acids Research <u>12</u>, 6159 (1984).
 b) L.H. Hurley, V.L. Reynolds, D.H. Swenson, G.L. Petzold, and T. A. Scahill, Science <u>226</u>, 843 (1984).
- a) W. Wierenga, J. Amer. Chem. Soc. <u>103</u>, 5621 (1981). b) P. Magnus and T. Gallagher, J. Chem. Soc., Chem. Commun. 389 (1984). c) V.H. Rawal and M.P. Cava, J. Chem. Soc., Chem. Commun. 1526 (1984). d) G.A. Kraus, S. Yue, and J. Sy, J. Org. Chem. <u>50</u>, 284 (1985). e) P. Magnus and S. Halazy, Tetrahedron Lett., 2985 (1985). f) R.E. Bolton, C.J. Moody, C.W. Rees, and G. Tojo, J. Chem. Soc., Chem. Commun. 1775 (1985). g) R.J. Sundberg and B.C. Pearce, J. Org. Chem. <u>50</u> 425 (1985). h) D.L. Boger and R.S. Coleman, J. Org. Chem., <u>49</u>, 2240 (1984). i) N. Komoto, Y. Enomoto, M. Miyagaki, X. Tanaka, K. Nitanai, and H. Umezawa Agric. Biol. Chem. <u>43</u>, 555 (1979). j) N. Komoto, Y. Enomoto, Y. Tanaka, K. Nitanai, and H. Umezawa, Agric. Biol. Chem. <u>43</u>, 559 (1979).
- 4. J.P. McGovren, G.L. Clarke, E.A. Pratt, and T.F. DeKoning. J. Antibiotics 37, 63 (1984).
- 5. M.A. Warpehoski, R.C. Kelly, J.P. McGovren, and W. Wierenga, Proc. Am. Assoc. Cancer Res. 26, 870 (1985).
- 6. P.G. Gassman, G. Gruetzmacher, and T.J. vanBergen, J. Amer. Chem. Soc. <u>96</u>, 5512 (1974) and references therein.
- 7. W. Wierenga, J. Griffen, and M.A. Warpehoski, Tetrahedron Lett., 2437 (1983).
- 8. W. Wierenga, unpublished results.
- 9. Sulfide 9b was prepared in 60% overall yield from ethyl-2-bromopropionate and mercaptoethanol by halide displacement, (K₂CO₃, acetone), saponification, acidification, and dehydration (toluene, reflux).
- 10. Alternatively, the mother liquors could be saponified and condensed with N-BOC-<u>D</u>-tryptophan. From this mixture the R:R diasteriomer could be fractionally recrystallized to >99% diasteriomeric purity.
- 11. 15 was prepared by condensation of indole-2-carboxylic acid and ethyl 5-amino-indole-2-carboxylate in DMF with ethyl dimethylaminopropylcarbodiimide (EDC), and saponification of the ester (50%). Ethyl 5-aminoindole-2carboxylate was obtained nearly quantitatively by catalytic hydrogenation of ethyl 5-nitroindole-2-carboxylate [S.M. Parmerter, A.G. Cook, and W.B. Dixon, J. Amer. Chem. Soc. <u>80</u>, 4621 (1958)].
- 12. ¹H^{*}NMR (DMSO-d₆) δ: 1.4 (m, 1H), 2.0 (s + m, 3H + 1H), 3.2 (m, 1H) 4.5 (m, 2H), 6.7 (s, 1H), 6.9 (s, 1H), 7.1-7.7 (m, 9H) 8.2 (brs, 1H), 10.3 (brs, 1H), 11.8 (brs, 1H), 11.9 (brs, 1H). [α]_D²⁵ + 155 (c = 0.278, DMF). The CD curve of U-71,184 patterned that of CC-1065, supporting its absolute stereochemistry as shown. ¹H-NMR for 16 (pyridine -d₅) δ: 13.2 (brs, 1H); 12.84 (brs, 1H); 12.07 (brs, 1H); 11.06 (brs, 1H); 8.93-8.64 (m, 2H + pyr.); 8.13 7.25 (m, 14 H + pyr); 5.3 (brs + m, 3H); 4.73 (m, 1H); 4.4-3.75 (m, 3H); 2.52 (brs, 3H). All new compounds were homogeneous by TLC and gave satisfactory ir, NMR, m.s., and exact mass and/or combustion analysis.
- 13. The technical assistance of J.L. Thompson, B.R. Evans, and A. Scott is gratefully acknowledged. (Received in USA 27 May 1986)